





15.	If D is the midpoint of the hypotenuse AC of isoscele $\triangle$ ABC, $\triangle$ ADB $\cong$ BDC according to co			celes right angled _ condition		
	A) ASA	B) SAS	C) SSS	D) RHS		
PH	YSICS					
16.	The horizontal c	The horizontal component of the weight of a body of mass m is				
	A) mg	B) $\frac{mg}{2}$	C) Zero	D) Infinity		
17.	Find the rectangular components (along X and Y axis) of a velocity					
	vector $10\sqrt{3}$ m/	s, making an ang	le of 60° to the X	- axis		
	A) 5√3 ,15	B) $5\sqrt{3}$ , 20	C) 10√3,20	D) 10√3 ,15		
18.	Vector $\vec{A}$ makes	equal angles with	x, y and z axis. V	alue of its compo-		
	nents (in terms o	f magnitude of $\vec{A}$	) will be			
	A) $\frac{A}{\sqrt{3}}$	B) $\frac{A}{\sqrt{2}}$	C) $\sqrt{3} A$	D) $\frac{\sqrt{3}}{A}$		
19.	How many mini	imum number of	coplanar vectors	s having different		
	magnitudes can	be added to give a	zero resultant			
	A) 2	B) 3	C) 4	D) 5		
20.	A particle moves along a circle with uniform speed V. When it has moved through an angle 60°, change in its velocity is					
	A) zero	B) $\sqrt{3}$ V	C) 3V	D) V		
21.	Which of the foll	owing set of force	s law keep an obj	ect in equilibrium		
	A) 5N, 6N, 1N	B) 5N , 4N, 7N	C) 1N, 2N, 5N	D) 5N, 6N, 20N		
22.	From the figure,		ī –	1		
	A) $\vec{A} + \vec{B} = \vec{C}$	B) $\vec{B} + \vec{C} = \vec{A}$		Ē		
	C) $\vec{C} + \vec{A} = \vec{B}$	D) $\vec{A} + \vec{B} + \vec{C} =$	<del>d</del>			



27.	$\vec{A} = 2\hat{i} + \hat{j}, B = 3\hat{j} - \hat{k}$ and $\vec{C} = 6\hat{i}$	$-2\hat{k}$ Value of $\vec{A} - 2\vec{B} + 3\vec{C}$ would be
	A) $20\hat{i} + 5\hat{j} + 4\hat{k}$	B) $20\hat{i} - 5\hat{j} - 4\hat{k}$
	C) $4\hat{i} + 5\hat{j} + 20\hat{k}$	D) $5\hat{i} + 4\hat{j} + 10\hat{k}$

28. A truck travelling due north at 20 *m/s* turns west and travels at the same speed. The change in its velocity be

A) 40 *m/s N*–*W* B) 
$$20\sqrt{2}$$
 *m/s N*–*W*

C) 40 m/s S–W D) 
$$20\sqrt{2}$$
 m/s S–W

29. As shown in figure the tension in the horizontal cord is 30 *N*. The weight *W* and tension in the string *OA* in Newton are



30. A body of 10 kg is suspended by a rope 50cm long and is pulled to a side through a horizontal distance of 30 cm by means of a horizontal force. Then that horizontal force is

A) 7.5 kg - wt B) 15 kg - wt C) 20 kg - wt D) 30 kg - wt

## **CHEMISTRY**

**31.** The number of moles of carbon dioxide which contain 8 g of oxygen is

A) 0.5 mol B) 0.20 mol C) 0.40 mol D) 0.25 mol

**32.** The total no of ions present in 111 g of  $CaCl_2$  is

A) One mole B) Two mole C) Three mole D) Four moles

## 33. Which has maximum number of atoms ?

A) 24g of c (12)	B) 56g of Fe (56)
C) 27g of Al (27)	D) 108g of Ag (108)

34.	One mole of oxygen gas is the volume of .				
	<ul><li>A) 1 litre of oxygen at S.T.P.</li><li>B) 32 litres of oxygen at S.T.P.</li><li>C) 22.4 litres of oxygen at S.T.P.</li></ul>				
	D) $6.02 \times 10^{23}$ molecules of oxygen at any temperature and pressu				
35.	Total number of atoms present 17 g of NH, is				
	A) $6.02 \times 10^{23}$ C) $3 \times 6.02 \times 110^{23}$		B) $2 \times 6.02 \times 10^{23}$		
			D) $4 \times 10^{-24}$ g	D) $4 \times 10^{-24}$ g	
36.	5.6 litres of oxygen at NTP is equivalent of				
	A) 1 mole	B) 1/2 mole	(C) 1/4 Mole	D) 1/8 mole	
37.	11.2 Litres of a gas at STP weighs 14g. The gas could not be :			ould not be :	
	A) <i>N</i> <sub>2</sub> <i>O</i>	B) CO	C) $B_2 H_6$	D) <i>N</i> <sub>2</sub>	
38.	Which one of	f the following state	tatements is incorect ?.		
	<ul> <li>A) One gram atom of carbon contains avogadro's number of atoms.</li> <li>B) One mole of oxygen gas contains Avogadro's number of atoms.</li> <li>C) One mole of hydrogen contains Avogadro's number of atoms.</li> <li>D) One mole of electrons stands for 6.02x 10<sup>23</sup> electrons</li> </ul>				
39.	One mole of atoms of oxygen represents				
	A) $6.02 \times 10^{23}$ atoms of oxygenB) $32$ gC) $22.4L$ of $O_2$ at STPD) $8$ g or		B) 32 g of oxyg	ofoxygen	
			D) 8g of oxygen		
40.	One mole of	sodium represents			
	A) 6.02×10 <sup>23</sup> atoms of sodium C) 11g of sodium		B) 46 gms of sodium		
			D) 34.5g of sod	D) 34.5g of sodium	
41.	The weight of 0.1 mole of $Na_2CO_3$ is				
	A) 106 g	B) 10.6 g	C) 5.3 g	D) 6.02×10 <sup>22</sup> g	
42.	The ratio bet and SO <sub>2</sub> is	ween the number	of molecules in eq	qual masses of $\operatorname{CH}_4$	
	A) 1:1	B) 4:1	C) 1:4	D) 2:1	

43.	What is the mole percentage of $\rm O_2$ in a mixture of 7g of $\rm N_2$ and 8g of $\rm O_2?$				
	A) 25%	B) 75%	C) 50%	D) 40%	
44.	Which of the foll conditions?	lowing gases has t	the highest densit	y under standard	
	A) CO	B) $N_2O$	C) C <sub>3</sub> H <sub>8</sub>	D) SO <sub>2</sub>	
45.	Which contains	more number of r	nolecules?		
	A) 1 mole of carb	on dioxide	B) 4g of hydrogen		
	C) 33.6 litres of o	xygen at STP	D) 6g of helium		
		MENTALA	BILITY		
Directions: In each of the following questions, a number series is given with one term missing. Choose the correct alternative that will continue the same pattern and replace the question mark in the given series.					
46.	1, 9, 25, 49, ?, 12	1, 9, 25, 49, ?, 121			
	(A) 64	(B) 81	(C) 91	(D) 100	
47.	4, 7, 12, 19, 28, 5	?			
	(A) 30	(B) 36	(C) 39	(D) 49	
48.	11, 13, 17, 19, 23	3, 25, ?			
	(A) 26	(B) 27	(C) 29	(D) 37	
49.	6, 12, 21, ?, 48				
	(A) 33	(B) 38	(C) 40	(D) 45	
50.	2, 5, 9, ?, 20, 27				
	(A) 14	(B) 16	(C) 18	(D) 24	

\*\*\*